## The Preparation of Ethyl 7-Azapteroate and Related Compounds (1)

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Sir:

The importance of the action of dihydrofolic reductase on the pyrazine ring of folic acid, especially the 7,8-double bond, and the structural relationship of the pteridine ring of folic acid to the 7-azapteridine (pyrimido [5,4-e]-astriazine) ring of toxoflavin (2) and related antibiotics (2,3) prompted us to investigate the preparation and ascertain the biological properties of 7-azafolic acid and its analogs. In this report the preparation of ethyl 7-azapteroate and its 4-amino and 4(3H)-thione analogs is described.

Alkylation of ethyl p-aminobenzoate with cyanomethyl p-toluenesulfonate (4) in ethyl acetate for 48 hours gave ethyl p-[(cyanomethyl)amino]benzoate (2) in 78% yield (5). The base catalyzed (6) addition of ethanol to the cyano group of 2 gave the imino ether 3 which was not isolated but condensed at room temperature with the

5-amino-4-hydrazinopyrimidine 1 (7) to give directly the 7-azapteridine 4(8) in 41.5% yield. Apparently a dihydro derivative of 4 is first formed, which undergoes air oxidation to give the heteroaromatic 7-azapteridine. Previously we demonstrated that the benzylthio group of this type of compound is readily displaced with nucleophiles (7). Treatment of an aqueous DMSO solution of 4 with potassium bicarbonate at 90° for 18 hours gave a 21% yield of ethyl 7-azapteroate (5) (8). Similarly, the reaction of 4 with a mixture of DMAc and 10% ethanolic ammonia at 60° for 5 hours gave the corresponding diamino-7azapteridine 6 (8) in 61% yield. The corresponding thione 7 (8) was obtained in 74% yield by reaction of 4 with hydrated sodium hydrosulfide in DMF at 80° for 2 hours. Several methods are being investigated for the incorporation of the remaining portion of the folic acid side chain.

 $R = \underline{p} - HNC_6H_4CO_2Et$ 

## REFERENCES

- (1) This work was supported by funds from the C. F. Kettering Foundation and the Chemotherapy, National Cancer Institute, National Institutes of Health Contract NIH-71-2021.
- (2) G. D. Daves, Jr., R. K. Robins, and C. C. Cheng, J. Am. Chem. Soc., 84, 1724 (1962).
- (3) E. C. Taylor and S. F. Martin, J. Org. Chem., 35, 3792 (1970).
- (4) S. Grudzinski, Acta Pol. Pharm., 24, 1 (1967); 24, 9 (1967).
- (5) D. H. Kim and R. L. McKee, J. Org. Chem., 35, 455 (1970).
  - (6) F. C. Schaefer and G. A. Peters, ibid., 26, 412 (1961).
- (7) C. Temple, Jr., C. L. Kussner, and J. A. Montgomery, *ibid.*, 36, 23 (1971).
- (8) Satisfactory elemental analyses and spectral data were obtained for this compound.